



Introduction to the protein models in Martini 3

Sebastian Thallmair

Martini Workshop 2025, Groningen

Today's Martini menu

Tuesday, August 12 Zernike campus, Bernoulliborg, building 5161

Room 5161.0151

09:00-10:30 hours: LECTURE: Martini model for proteins

Sebastian Thallmair/Adolfo Poma

10:30-11:00 hours: Tea/Coffee break

11:00-12:30 hours: LECTURE: Martini IDPs/Highres backbone/ProLint

Chris Brasnett/Luis Borges/Daniel Ramirez

12:30-13:30 hours: LUNCH (bring your own)

Zernike campus, Linnaeusborg, building 5173

Room 5173.0169 (Tea/coffee at around 3 PM)

13:30-17:00 hours: HANDS-ON SESSION 2: MARTINI PROTEINS

Tutorials at different levels on proteins

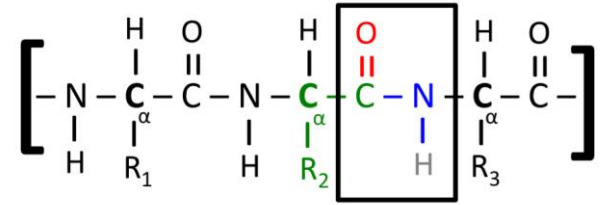
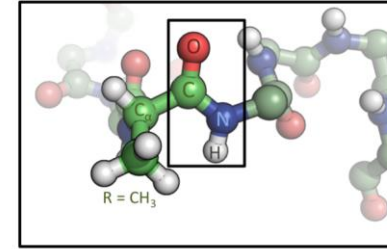
Zernike campus, area in front of building 5161.0151

17:00-19:00 hours: POSTER SESSION with SNACKS & DRINKS

A list of titles will be provided.

Proteins

- Essential nutrients for the human body – 17 kJ/g
lipids 33 kJ/g
- Biopolymers and macromolecules
consisting of 20 natural amino acids
- Large spectrum of functions
 - catalysis
 - signaling
 - providing structure
 - transport



<https://en.wikipedia.org/wiki/Protein>



John Kendrew, 1958



AlphaFold, 2025

Coarse Grained Model for Semiquantitative Lipid Simulations

Siewert J. Marrink,^{*} Alex H. de Vries, and Alan E. Mark

Department of Biophysical Chemistry, University of Groningen, Nijenborgh 4, 9747 AG Groningen, The Netherlands

Received: August 22, 2003; In Final Form: October 29, 2003

7812

J. Phys. Chem. B **2007**, 111, 7812–7824

The MARTINI Force Field: Coarse Grained Model for Biomolecular Simulations

Siewert J. Marrink,^{*,†} H. Jelger Risselada,[†] Serge Yefimov,[‡] D. Peter Tieleman,[§] and Alex H. de Vries[†]

Groningen Biomolecular Sciences and Biotechnology Institute & Zernike Institute for Advanced Materials, Department of Biophysical Chemistry, University of Groningen, Nijenborgh 4, 9747 AG Groningen, The Netherlands, Zernike Institute for Advanced Materials, Department of Applied Physics, University of Groningen, Nijenborgh 4, 9747 AG Groningen, The Netherlands, and Department of Biological Sciences, University of Calgary, 2500 University Drive NW, Calgary AB T2N 1N4, Canada

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ARTICLES

<https://doi.org/10.1038/s41592-021-01098-3>

nature | methods

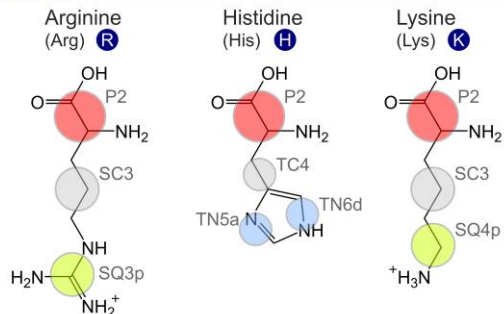
Check for updates

Martini 3: a general purpose force field for coarse-grained molecular dynamics

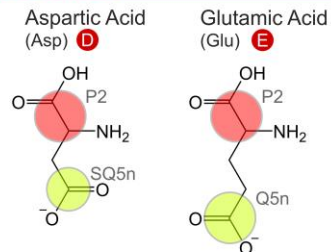
Paulo C. T. Souza^{1,2}, Riccardo Alessandri¹, Jonathan Barnoud^{1,3}, Sebastian Thallmair^{1,4}, Ignacio Faustino¹, Fabian Grünewald¹, Ilias Patmanidis¹, Haleh Abdizadeh¹, Bart M. H. Bruininks¹, Tsjerk A. Wassenaar¹, Peter C. Kroon¹, Josef Melcr¹, Vincent Nieto², Valentina Corradi⁵, Hanif M. Khan^{5,6}, Jan Domański^{7,8}, Matti Javanainen^{9,10}, Hector Martinez-Seara⁹, Nathalie Reuter⁶, Robert B. Best⁸, Ilpo Vattulainen^{10,11}, Luca Monticelli¹², Xavier Periole¹², D. Peter Tieleman⁵, Alex H. de Vries¹ and Siewert J. Marrink¹

CG Martini 3 representations of amino acids

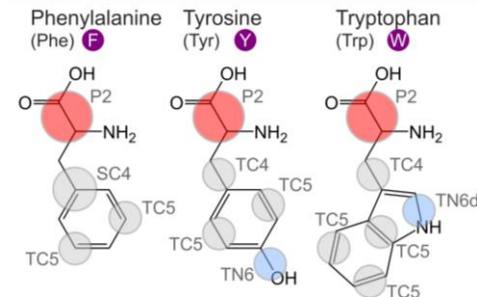
Basic Residues*



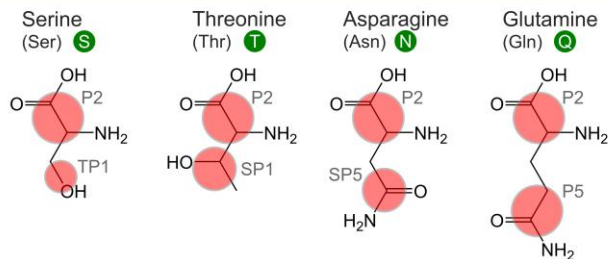
Acidic residues*



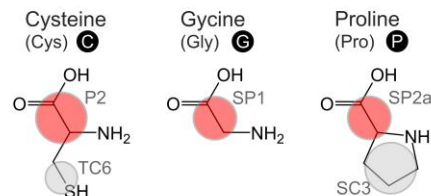
Aromatic residues



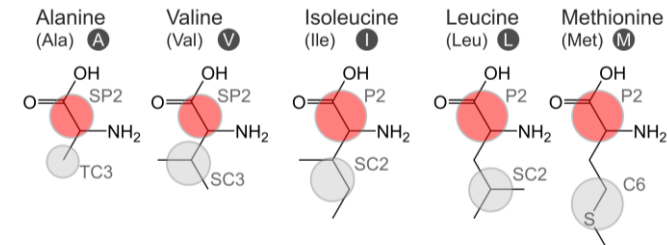
Polar Residues



Special case residues



Non-aromatic hydrophobic residues



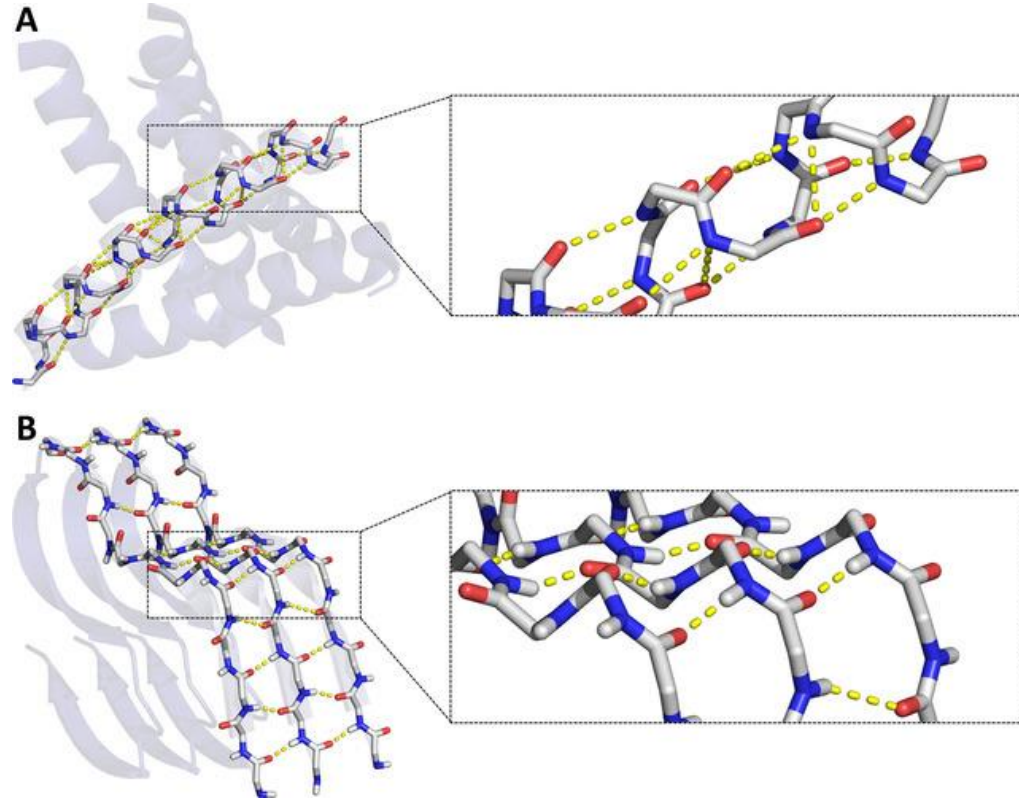
* Most common protonation state is shown

[1] T. Duve, *et al.*, *bioRxiv* (2025).

[2] P. C. T. Souza, *et al.*, *Nat. Commun.* **16**, 4051 (2025).

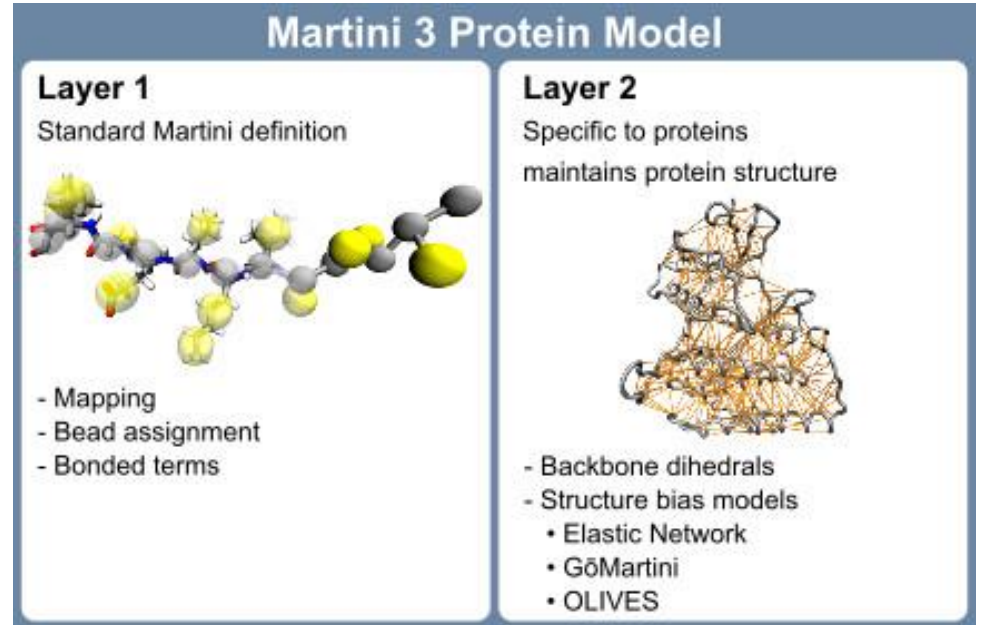
Protein structures and their key interactions

- Different types of interactions stabilize proteins:
 - hydrophobic interactions
 - polar interactions
 - hydrogen bonds
 - charged interactions
 - disulfide bonds, etc.
- In particular the directionality of H bonds is challenging in CG models
- Additional bonded interactions are needed



CG Martini 3 protein model has two layers

- Layer 1: standard Martini definition
- Layer 2: secondary structure-specific dihedrals and structure bias models
- Both layers are decoupled and can be developed independently

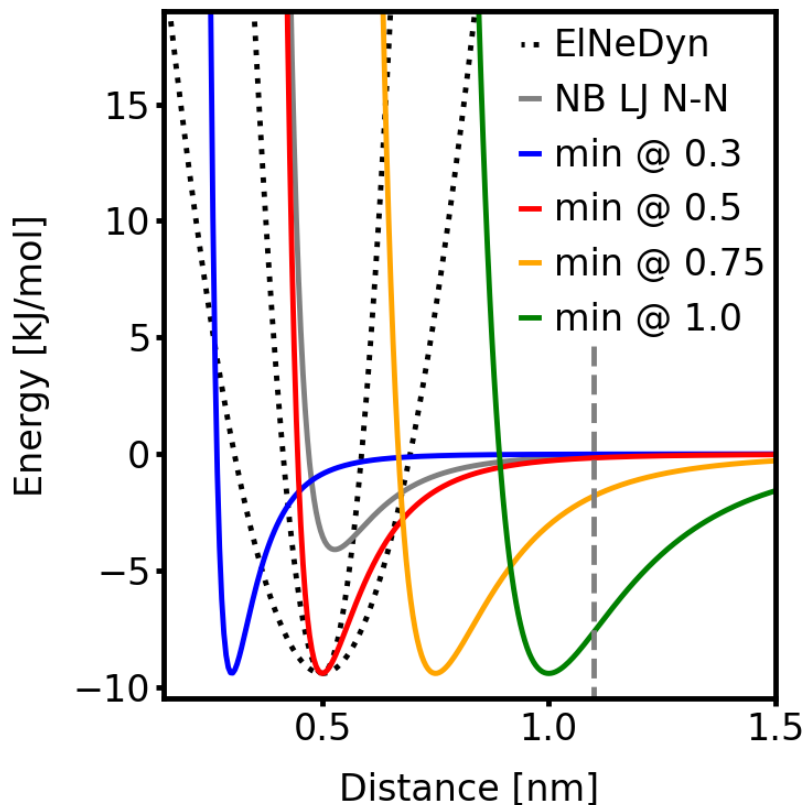


[1] T. Duve, *et al.*, *bioRxiv* (2025).

[2] P. C. T. Souza, *et al.*, *Nat. Commun.* **16**, 4051 (2025).

Additional stabilization required for tertiary structure

- Currently, three choices for structure bias models:
 - **Elastic Network** (EN) with harmonic potentials [3,4]
 - **GōMartini 3**: Gō-like model with Lennard-Jones (LJ) potentials [2]
 - **OLIVES**: Gō-like model with Lennard-Jones (LJ) potentials [5]
- Proteins part I(b) tutorial



[2] P. C. T. Souza, *et al.*, *Nat. Commun.* **16**, 4051 (2025).

[3] P. C. T. Souza, *et al.*, *Nat. Methods* **18**, 382 (2021).

[4] P. C. Kroon, *et al.*, *eLife* **12**, PR90627 (2023).

[5] K. B. Pedersen, *et al.*, *J. Chem. Theory Comput.* **20**, 8049 (2023).

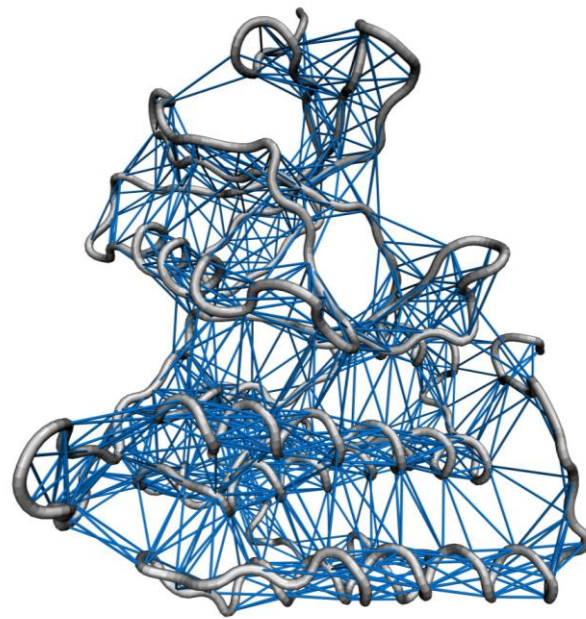
Elastic network model

- Bonded terms applied to backbone (BB) beads at the backbone center of mass
- Elastic bond introduced if:
 - residues are separated by at least 2 residues
 - BB distance d is shorter than the cutoff distance R_c
- Harmonic potential connects the BB beads

$$V(d) = 0.5 k (d - R_{eq})^2$$

- Typical setup:
 - cutoff distance $R_c = 0.8 - 1.0$ nm
 - force constant $k = 700 - 1000$ kJ/(mol nm²)

Elastic Network



Preparing and fine-tuning your elastic network model

- Martinize2 command [4]:

```
martinize2 <...> -ff martini3001 -elastic -ef 700.0 -el 0.5 -eu 0.9
```

- Optimization of
cutoff distance R_c and
force constant k

- Additional fine-tuning options:

- distance-dependent force constant k
options `-ea -ep`
- Removal of EN bonds in flexible
protein regions

```
[ bonds ]
```

```
...
```

```
; Rubber band
```

```
1      7  1  0.79915  700.0
```

```
1     11  1  0.6977  700.0
```

```
1     13  1  0.8159  700.0
```

```
1    364  1  0.7891  700.0
```

```
3     11  1  0.47668  700.0
```

```
3     13  1  0.57024  700.0
```

```
3     15  1  0.81049  700.0
```

```
3    212  1  0.85239  700.0
```

```
...
```

[4] P. C. Kroon, *et al.*, *eLife* **12**, PR90627 (2023).
www.github.com/marrink-lab/vermouth-martinize

GōMartini 3 model

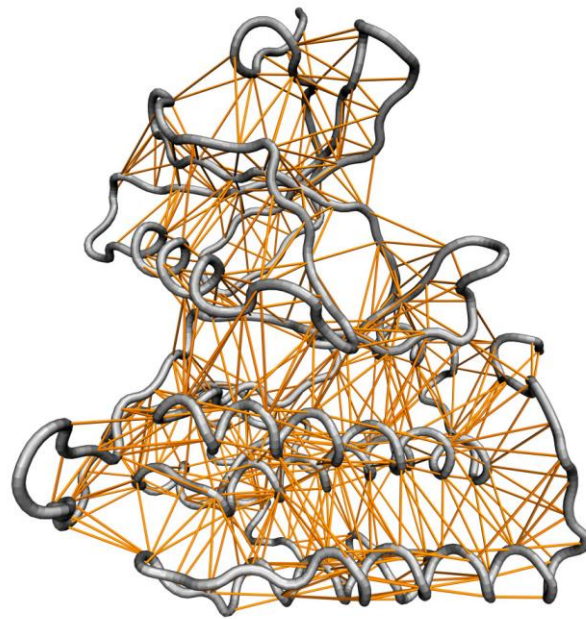
- Bonded terms applied to backbone (BB) beads at the backbone center of mass
- Native contacts define the introduced bonds
- Gō bond introduced if:
 - residues are separated by at least 3 residues
 - BB distance d is shorter than the cutoff distance R_c
- LJ potential connects the BB beads

$$V_{LJ}(d) = 4\varepsilon \left[\left(\frac{\sigma}{d} \right)^{12} - \left(\frac{\sigma}{d} \right)^6 \right]$$

- Typical setup:

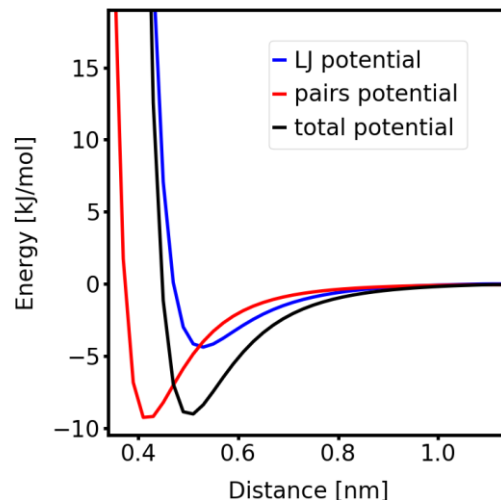
cutoff distance	$R_c = 1.0 - 1.1$ nm
dissociation energy	$\varepsilon = 9.4 - 15$ kJ/mol

GōMartini



Implementation of the GōMartini 3 model

- Virtual sites are used for the Gō-like interactions [5]
 - Generation of additional bead types
 - Non-bonded cutoff is applied to Gō-like interactions
→ advantageous for domain decomposition
- Regular non-bonded terms excluded for Gō-like contacts
→ more accurate protein structures
- Contact map based on
 - Overlap criterion of all amino acid atoms
 - Additional contacts via repulsive contacts of structural units (rCSU) [6]



[2] P. C. T. Souza, *et al.*, *Nat. Commun.* **16**, 4051 (2025).

[6] K. Wótek, *et al.*, *J. Chem. Phys.* **143**, 243105 (2015).

Preparing and fine-tuning your Gō-like model

- Files generated by Martinize2 (or `create_goVirt.py`):

`go_atomtypes.itp`

`go_nbparams.itp`

virtual bead definitions

Gō interaction table

- Definition of `GO_VIRT` variable activates Gō-like model

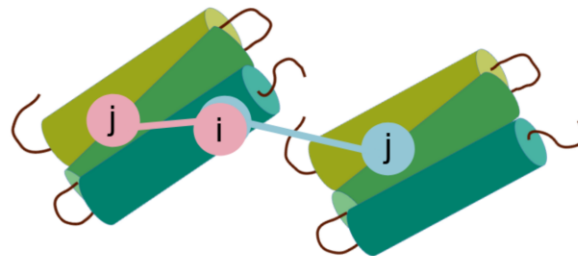
- Files have to be included in the `martini.itp` file

- Pitfalls:

- Multiple chains / protein copies
- Atomistic `.pdb` file

OliGōmers: Extension of GōMartini 3

- Addresses the challenge of simulating oligomers with GōMartini 3
- Multi-layered virtual site implementation
 - intra pairs
 - inter pairs
 - regular CG-BB



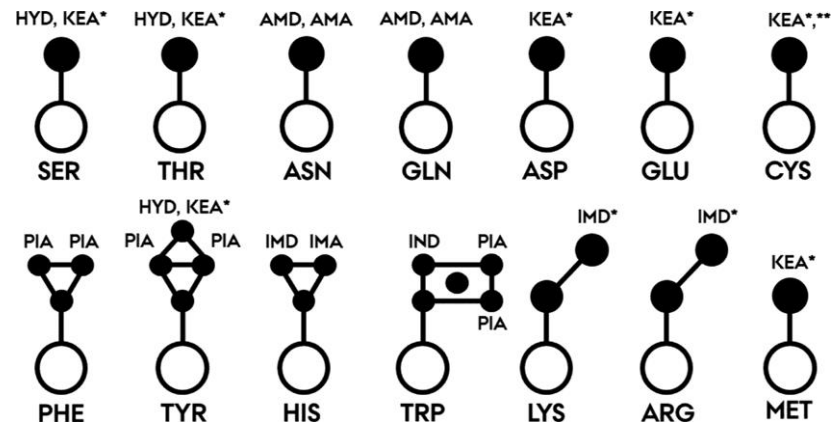
INTRA Gō i-j pair			INTER Gō i-j pair		
	intra	inter		intra	inter
INTRA Gō	VS _{intra} (ϵ_{intra})	•	VS _{inter} (ϵ_{inter})	•	•
	VS _{inv} ($\epsilon_{inv} = -\epsilon_{intra}$)	•	Excl. inter	•	
	Excl. inv	•	CG-BB (ϵ_{BB})	•	•
INTER CG-BB	CG-BB (ϵ_{BB})	•	VS _{inv-BB} ($\epsilon_{inv-BB} = -\epsilon_{BB}$)	•	•
	Excl. CG-BB	•	Excl. D	•	

OLIVES model

- Bonded terms applied to the beads containing the interacting atoms/fragments
- Eight classes define the introduced bonds
- LJ potential connects the BB beads

$$V_{LJ}(d) = 4\varepsilon \left[\left(\frac{\sigma}{d} \right)^{12} - \left(\frac{\sigma}{d} \right)^6 \right]$$

- QM calculated H bond dissociation energies
 $\varepsilon = 8.8 - 33.3 \text{ kJ/mol}$
 \rightarrow scaling option is available (but not tested)



[kcal/mol]	HYD	AMD	IMD	IND
KEA	3.54	4.06	5.78	5.12
AMA	4.51	5.44	6.92	6.27
IMA	6.24	6.55	7.96	7.11
PIA*	4.97**	5.31	6.17	2.10

^a(HYD) hydroxyl donor, (AMD) amide donor, (IMD) imidazole donor, (IND) indole donor, (KEA) ketone donor, (AMA) amide acceptor, (IMA) imidazole acceptor, and (PIA) π acceptor. Note that the majority of the results were obtained by DFT calculations, which have a typical error of 2 kcal/mol.³⁴ * π acceptor energies are divided by 2 internally, spreading the energy across two beads, as illustrated in Figure 1. **Average of two results in the original work by Du et al.¹⁸

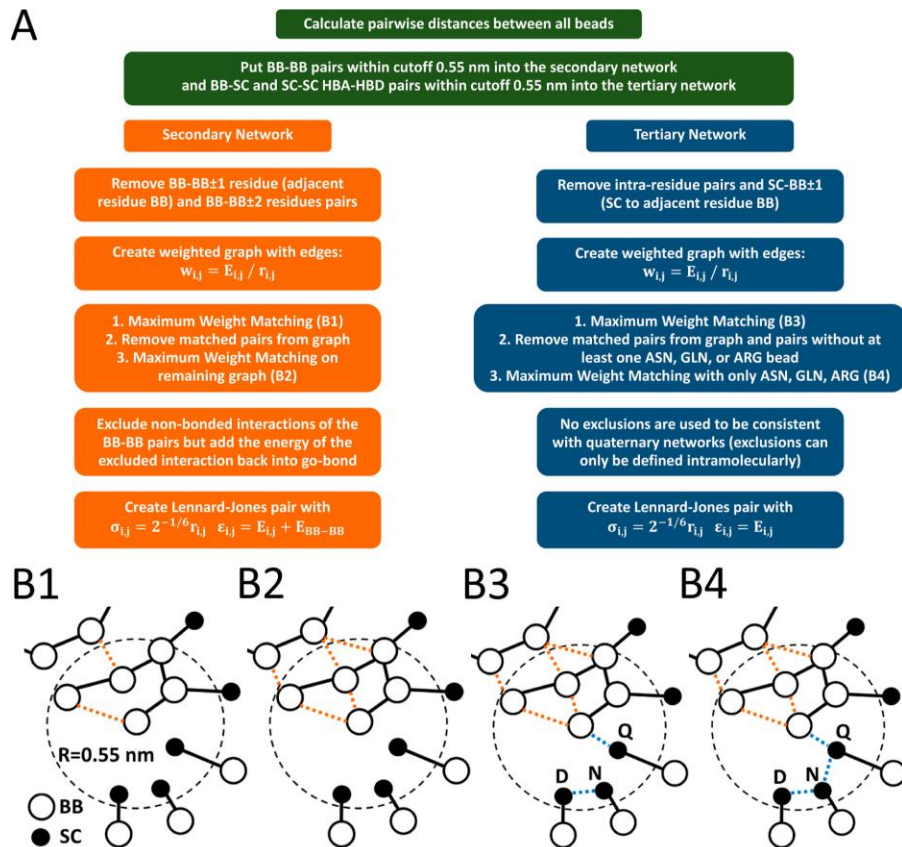
OLIVES model

- Bonded terms applied to the beads containing the interacting atoms/fragments
- Eight classes define the introduced bonds
- LJ potential connects the BB beads

$$V_{LJ}(d) = 4\epsilon \left[\left(\frac{\sigma}{d} \right)^{12} - \left(\frac{\sigma}{d} \right)^6 \right]$$

- QM calculated H bond dissociation energies
 $\epsilon = 8.8 - 33.3 \text{ kJ/mol}$
 \rightarrow scaling option is available (but not tested)
- Gō bond introduced if:
 - BB-BB: separated by at least 2 residues
 - BB-SC: separated by at least 1 residue
 - distance d is shorter than the cutoff $R_c = 0.55 \text{ nm}$

A

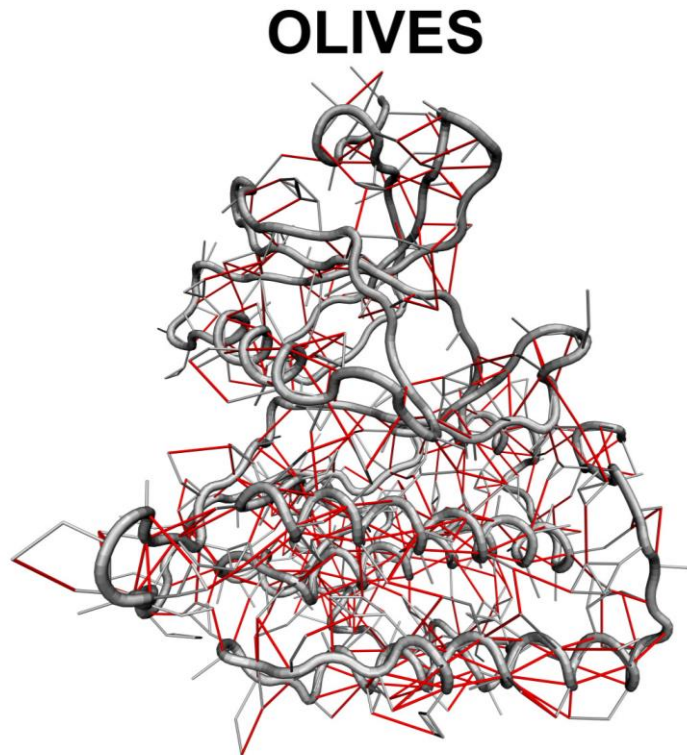


OLIVES model

- Bonded terms applied to the beads containing the interacting atoms/fragments
- Eight classes define the introduced bonds
- LJ potential connects the BB beads

$$V_{LJ}(d) = 4\varepsilon \left[\left(\frac{\sigma}{d} \right)^{12} - \left(\frac{\sigma}{d} \right)^6 \right]$$

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- Gō bond introduced if:
 - BB-BB: separated by at least 2 residues
 - BB-SC: separated by at least 1 residue
 - distance d is shorter than the cutoff $R_c = 0.55$ nm



Comparison of the different protein models

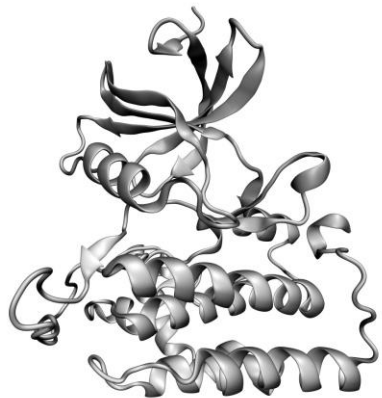
Martini w/o bias
no additional bonds

Elastic Network
BB distances define
contact map
harmonic potentials
1004 bonds

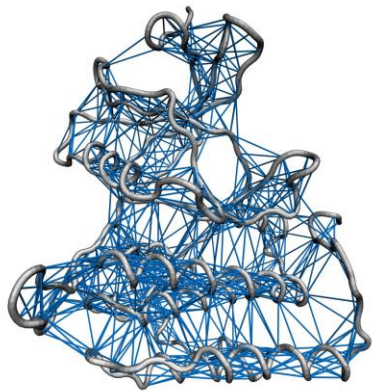
GōMartini 3 model
natural contacts define
contact map
LJ potentials
609 bonds

OLIVES model
chemical contacts
define contact map
LJ potentials
374 bonds

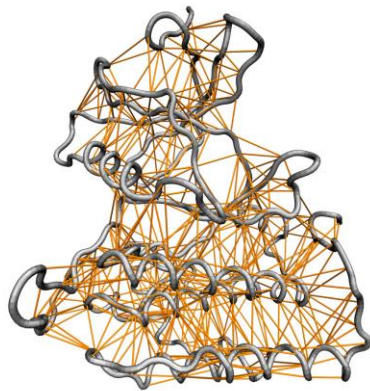
Atomistic



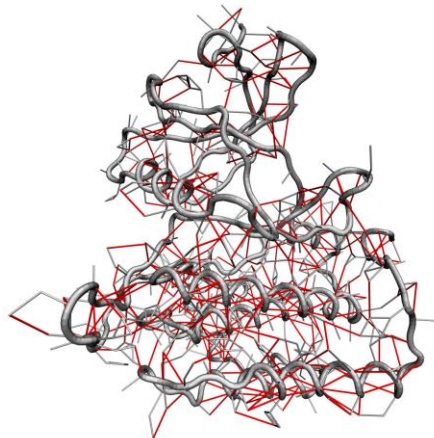
Elastic Network



GōMartini

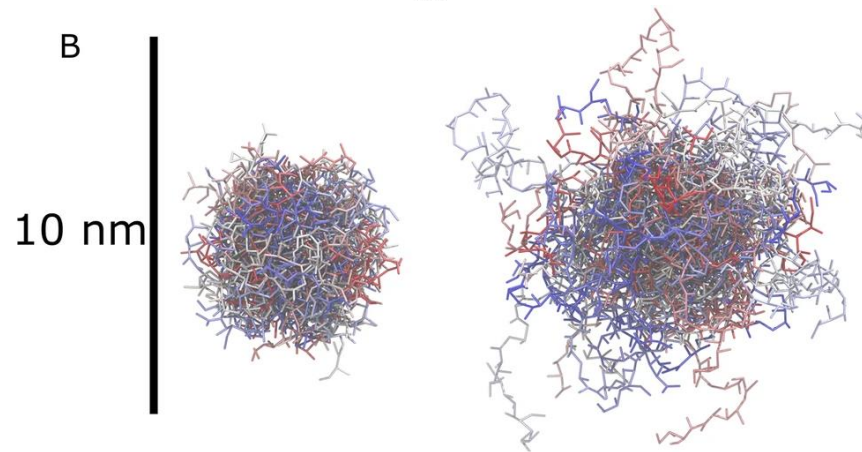
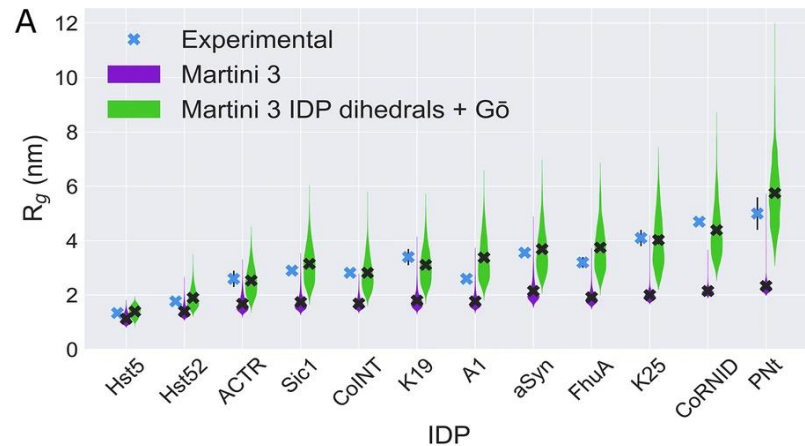


OLIVES



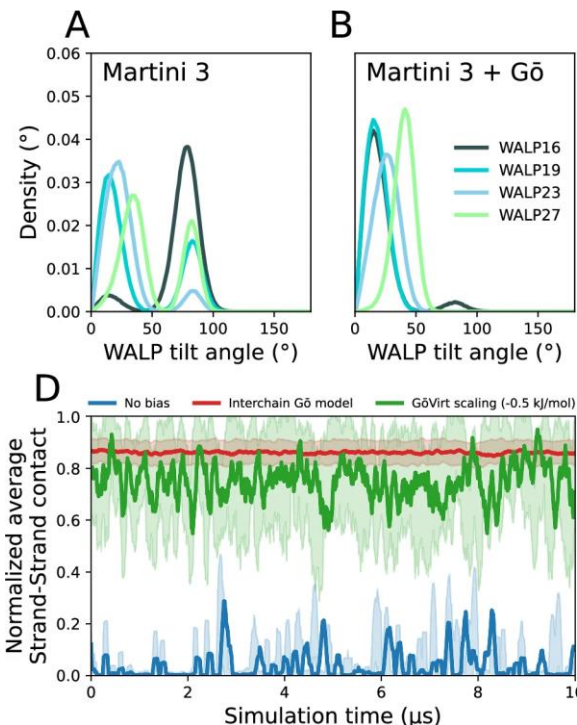
Protein models for intrinsically disordered proteins (IDPs)

- Water bias via virtual BB sites of GōMartini 3
- Flexible approach for environmental bias corrections
 - IDPs



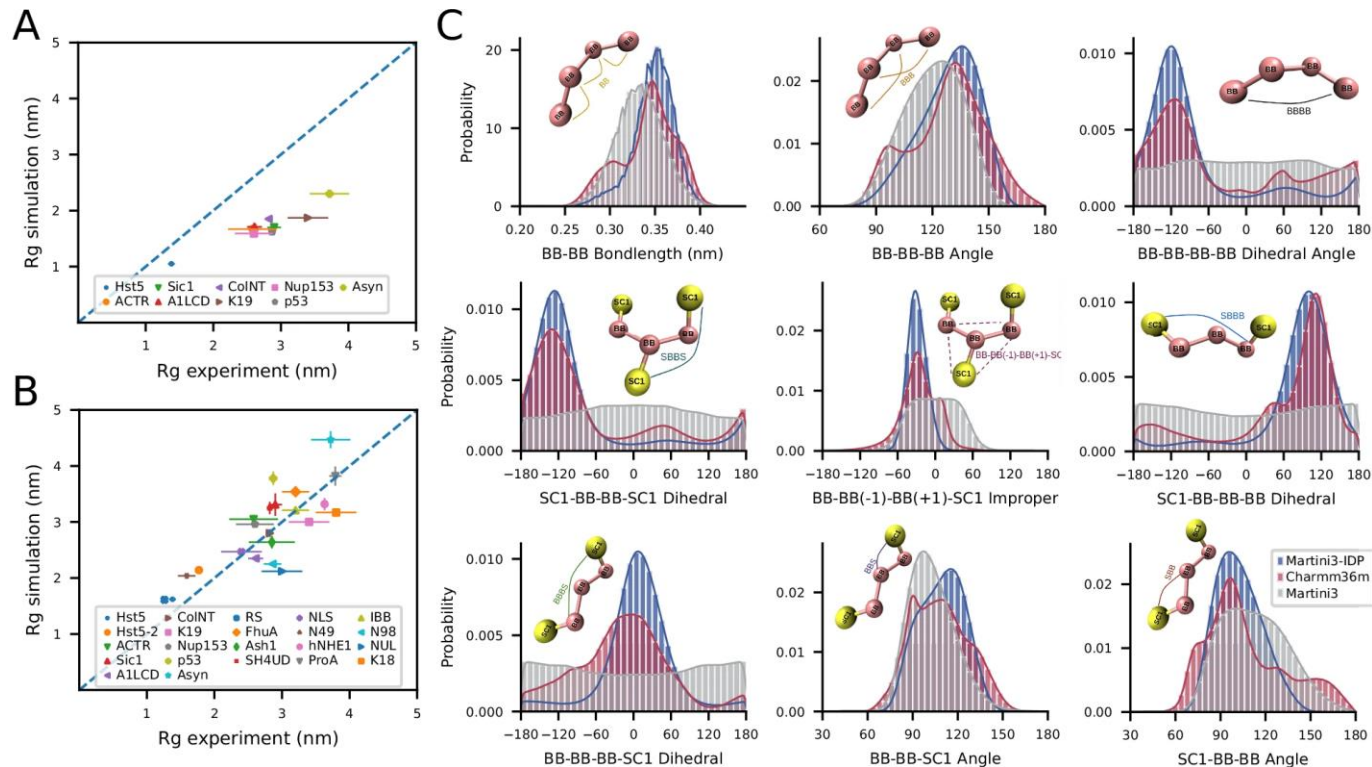
Protein models for intrinsically disordered proteins (IDPs)

- Water bias via virtual BB sites of GōMartini 3
- Flexible approach for environmental bias corrections
 - IDPs
 - single-pass transmembrane helices
 - β -sheet dimers



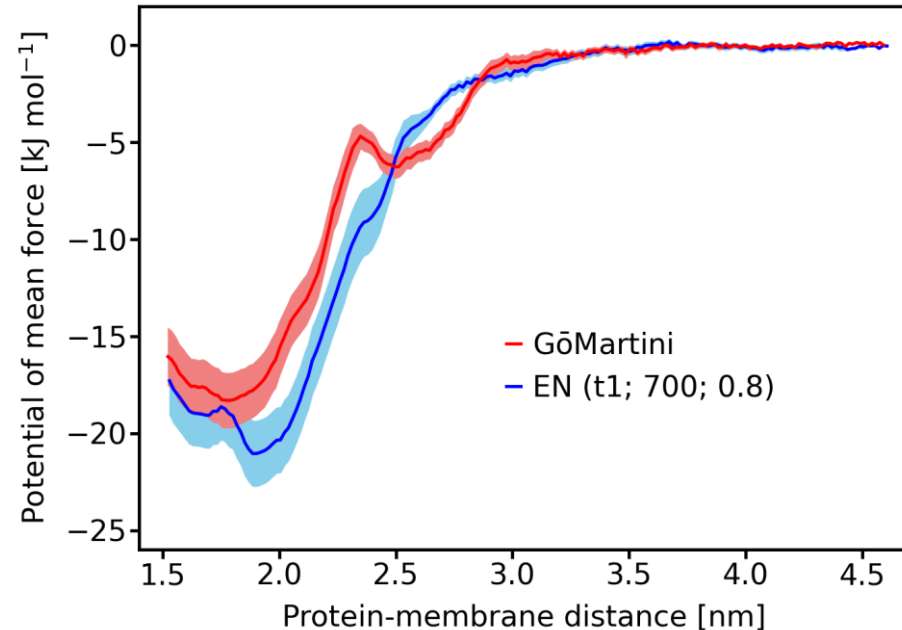
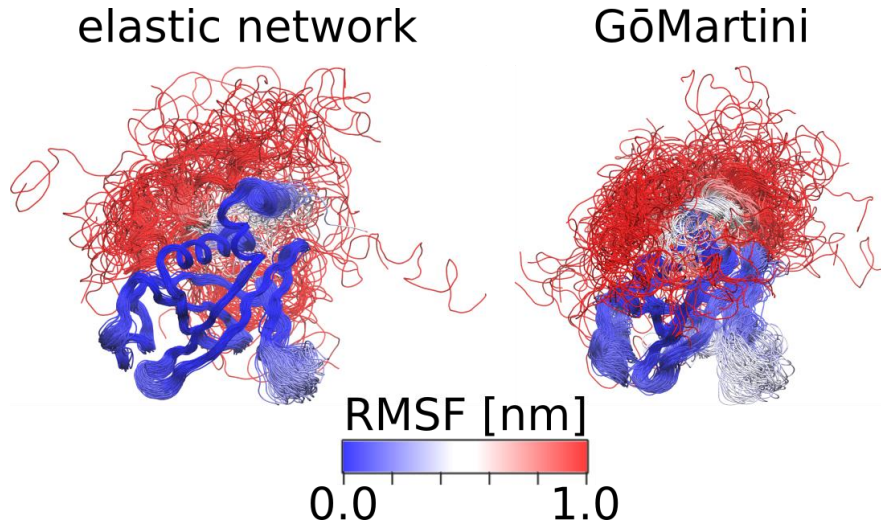
Protein models for intrinsically disordered proteins (IDPs)

- Martini-IDP:
a specific Martini protein model for IDPs
- Dihedral angels in the backbone added
- Improve the too high compactness of IDPs
- Proteins part I(c) tutorial



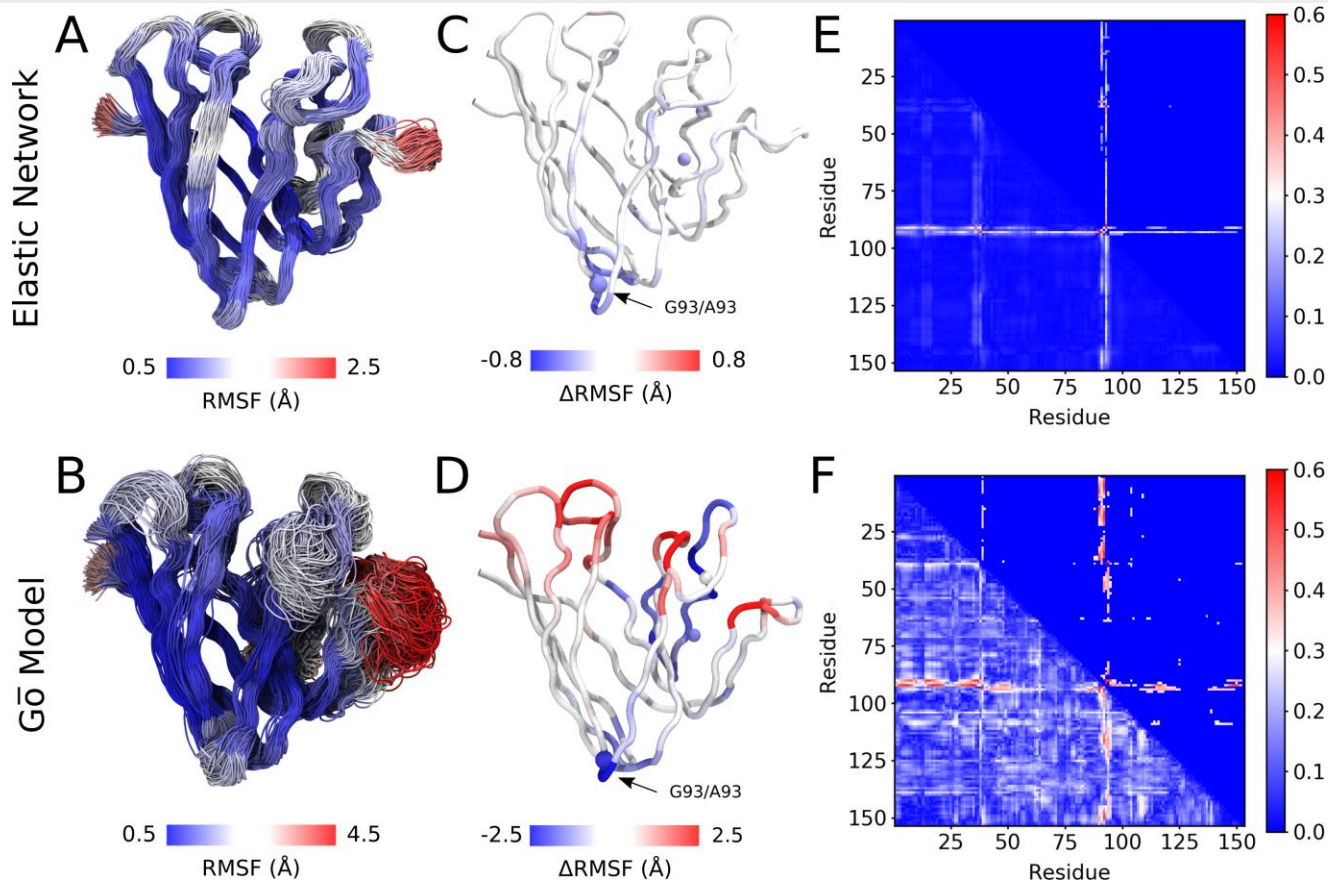
Show case I – PH domains

- PI(4,5)P₂ binding PH domain of phospholipase C ($\delta 1$)
- Elastic network more rigid than GōMartini
- GōMartini suggests long-distance minimum known from other PH domains [7]



Show case II – allosteric pathway in SOD1 mutant

- G93A mutant destabilizes electrostatic loop
- Increased spatial demand of the mutation
- Elastic network results in highly rigid protein model
- GōMartini model enables structural response of the protein



Summary

- Directionality of **hydrogen bonds** is **important** for protein structures
- **Faithful representation** of tertiary and quaternary structure of proteins **requires additional layer of stabilizing interactions**
- **Elastic network:**
 - Harmonic potentials from distance criterion
 - Rigid protein structure
- **GōMartini:**
 - Lennard-Jones potentials from native contact map
 - More flexible protein structure
- **OLIVES:**
 - Lennard-Jones potentials from H bonding network
 - More flexible protein structure
- Intrinsically disordered proteins:
water-bias corrections or Martini3-IDP

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Luca Monticelli
Tim Carpenter



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Jannet Nijhuis-Kampen

... and all contributors and attendees to the
Martini Workshop 2025

